

4-[[*E*-(3,5-Dimethyl-1-phenyl-1*H*-pyrazol-4-yl)methylidene]amino]-1,5-dimethyl-2-phenyl-1*H*-pyrazol-3(2*H*)-one

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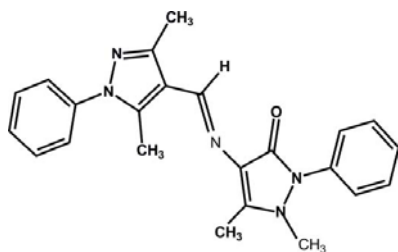
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Key indicators: single-crystal X-ray study; *T* = 296 K; mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$; *R* factor = 0.059; *wR* factor = 0.161; data-to-parameter ratio = 19.3.

The title Schiff base compound, $\text{C}_{23}\text{H}_{23}\text{N}_5\text{O}$, was synthesized by the reaction of 4-aminophenazone and 3,5-dimethyl-1-phenylpyrazole-4-carboxaldehyde. The molecule adopts an *E* configuration about the central $\text{C}=\text{N}$ double bond. A weak intramolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bond generates an *S*(6) ring motif. The dihedral angle between the pyrazole rings is $24.72(10)^\circ$ and the dihedral angles between the pyrazole rings and the adjacent phenyl rings are $58.67(10)$ and $46.58(11)^\circ$. The crystal structure is stabilized by weak $\text{C}-\text{H}\cdots\pi$ interactions involving the pyrazolone and phenyl rings.

Related literature

For background to and applications of heterocyclic Schiff bases, see: Nawaz *et al.* (2009); Li *et al.* (1999); Urena *et al.* (2003); Geronikaki *et al.* (2003); Shanker *et al.* (2009); Pandeya *et al.* (1999); Sridhar *et al.* (2002); Nawrocka *et al.* (2004). For related structures, see: Eryigit & Kendi (1998); Manikandan *et al.* (2000). For details of hydrogen-bond motifs, see: Bernstein *et al.* (1995).



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Experimental

Crystal data

$\text{C}_{23}\text{H}_{23}\text{N}_5\text{O}$
M_r = 385.46
Monoclinic, $P2_1/c$
a = 15.2985 (2) \AA
b = 7.6827 (1) \AA
c = 19.6737 (3) \AA
 β = 116.905 (1) $^\circ$

V = 2062.03 (5) \AA^3
Z = 4
Mo *K* α radiation
 μ = 0.08 mm^{-1}
T = 296 K
0.45 \times 0.21 \times 0.10 mm

Data collection

Bruker SMART APEXII CCD
area-detector diffractometer
Absorption correction: multi-scan
(*SADABS*; Bruker, 2009)
T_{min} = 0.965, *T_{max}* = 0.992

23014 measured reflections
5993 independent reflections
2881 reflections with $I > 2\sigma(I)$
R_{int} = 0.048

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.059$
 $wR(F^2) = 0.161$
S = 1.03
5993 reflections
311 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.24 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.18 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

*Cg*1 and *Cg*2 are the centroids of the N4/N5/C11–C13 and C1–C6 rings, respectively.

| <i>D</i> – <i>H</i> ⋯ <i>A</i> | <i>D</i> – <i>H</i> | <i>H</i> ⋯ <i>A</i> | <i>D</i> ⋯ <i>A</i> | <i>D</i> – <i>H</i> ⋯ <i>A</i> |
|--------------------------------------|---------------------|---------------------|---------------------|--------------------------------|
| C10–H10A⋯O1 | 0.986 (18) | 2.40 (2) | 3.052 (3) | 123.3 (14) |
| C19–H19A⋯ <i>Cg</i> 2 ⁱ | 0.990 (19) | 2.656 (19) | 3.452 (2) | 137.4 (17) |
| C20–H20C⋯ <i>Cg</i> 1 ⁱⁱ | 0.96 | 2.85 (3) | 3.720 (3) | 149 (1) |
| C22–H22B⋯ <i>Cg</i> 2 ⁱⁱⁱ | 0.96 | 2.82 (3) | 3.585 (3) | 135 (1) |

Symmetry codes: (i) $-x + 1, -y - 1, -z - 1$; (ii) $-x, -y - 2, -z - 1$; (iii) $-x + 1, y - \frac{1}{2}, -z - \frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH5060).

References

- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
Bruker (2009). *APEX2*, *SAINT* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
Eryigit, R. & Kendi, E. (1998). *J. Chem. Crystallogr.* **28**, 145–147.
Geronikaki, J. M. A., Litina, D. H. & Amourgiannou, M. (2003). *Farmaco*, **58**, 489–495.